

## Novichem Co.

Your Chemical Partner

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# HYDROGEN PEROXIDE (H2O2)

#### **DESCRIPTION**

**Hydrogen peroxide** is formulated with an inorganic tin-based stabilizer system for high stability and long-term storage. The 70% concentration can be diluted to 35% or 50% concentration with most ordinary tap water found. This permits a customer to achieve maximum economy by purchasing hydrogen peroxide at 70% concentration and diluting it to 35% or 50% for storage.

#### **TECHNICAL SPECIFICATIONS**

	35%	<b>50%</b>	70%
Loss in % assay, 1 yr, 25°C	<0.7	<1.0	<1.4
Apparent pH	<3.7	<3.0	0-1.0
Sn, ppm	25-35	36-50	50-70
Active oxygen content, %	16.5	23.5	32.9
Specific gravity (20°C/4°C)	1.13	1.20	1.29
Lbs/gal (kg/M <sup>3</sup> or g/L) @ 20°C	9.4 (1126)	10.0 (1198)	10.8 (1294)
Boiling point, °C (°F)	108 (226)	114 (237)	126 (259)
Freezing point, °C (°F)	-33 (-27)	-52 ( <del>-</del> 62)	-40 (-40)

Packing: 35 Kg PE Drums

### **APPLICATIONS**

Among the oxidizing bleaching agents, only hydrogen peroxide provides a high bleaching effect at reasonable costs, especially if modern short term bleaching processes are used with only a few minutes bleachingtime.

- Peroxide bleaching keeps the fibre quality.
- Cotton can be bleached with peroxide in a single stage. Other processes require two or three bleaching stages.
- Hot, alkaline, bleaching has not only a bleaching, but also a cleaning effect, it therefore combines theadvan tages of alkaline traction with the bleaching treatment.
- Animal fibres can only be bleached with peroxide to a high and stable degree of whiteness. corrosion of stainless steel equipment does not occur during peroxide bleaching.

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